

Sec A

Ans 1: System - It is defined as a quantity of matter or a region in space upon which attention is concentrated in the analysis of a problem.

Surrounding  $\Rightarrow$  Everything out of the system is called surrounding.

Boundary: The system is separated from the surrounding by the system boundary.

Ans 2: Clausius statement: It is impossible to construct a device which operating in a cycle, will produce no effect other than transfer of heat from a cooler to a hotter body.

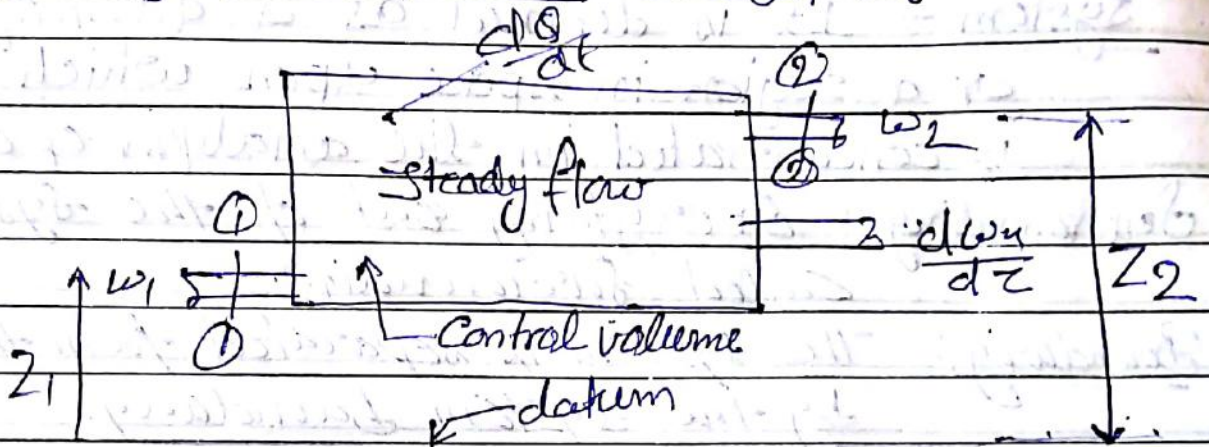
Ans 3: Extensive properties: - which properties depend on the mass called extensive  
Intensive properties: - which properties independent on the mass called intensive

Q. 5: Steady state process: - Any thermodynamic property will have a fixed value at a particular location, and will not alter with time.

Sec B



Ans. 1 Steady flow process energy eq<sup>n</sup> →



$$h_1 + \frac{v_1^2}{2} + z_1 g + \frac{dQ}{dm} = h_2 + \frac{v_2^2}{2} + z_2 g + \frac{dW}{dm}$$

Q.2

$$T_1 = 273 + 25 = 298$$

$$T_2 = 273 + (-20) = 253$$

Power requirement

$$\frac{Q_2}{T_2} = \frac{Q_1}{T_1}$$

$$Q_1 = \frac{333 - 33}{253} \times 298 = 392.16 \text{ kJ/t}$$

$$W = Q_1 - Q_2 = 392 - 333 = 59 \text{ kW}$$

273+25=298

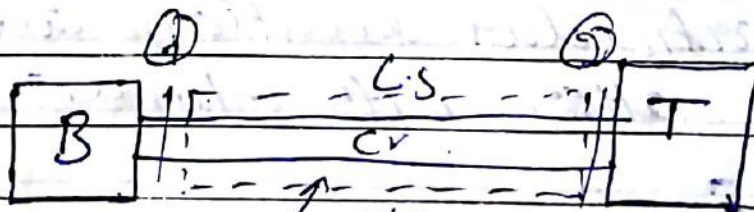


20,000 kJ/h

253

273-20

Ans 3-5



$$\frac{dQ}{dm} = -8.5 \text{ kJ/kg}$$

$$\frac{A_1 V_1}{v_1} = \frac{A_2 V_2}{v_2}$$

$$V_2 = \frac{A_1 V_1}{v_1} \cdot \frac{v_2}{A_2} \cdot V_1 = \frac{0.084}{0.073} V_1 = 1.15 V_1$$

$$\frac{dW_{out}}{dm} = 0$$

$$h_1 + \frac{V_1^2}{2} + \frac{dQ}{dm} = h_2 + \frac{V_2^2}{2}$$

$$\frac{(V_2^2 - V_1^2) \times 10^{-3}}{2} = h_1 - h_2 + \frac{dQ}{dm}$$

$$= 3213.6 - 3202.6 + (-8.5) = 2.5 \text{ kJ/kg}$$

$$V_1^2 (1.15^2 - 1^2) = 5 \times 10^3$$

$$V_1^2 = 15,650 \text{ m}^2/\text{s}^2$$

$$V_1 = 125.1 \text{ m/s}$$

$$\therefore \text{mass flow rate } \dot{m} = \frac{A_1 V_1}{v_1} = \frac{\pi}{4} (0.2)^2 \text{ m}^2 \times 125.1 \text{ m/s} / 0.073 \text{ m}^3/\text{kg}$$

$$= 53.8 \text{ kg/s}$$

$$\eta_{th} = \frac{\text{work output}}{\text{heat supplied}} = \frac{W}{Q_1}$$

$$W = 10 \text{ kW} = 10 \text{ kJ/s} = 600 \text{ kJ/min}, \quad Q_1 = 2250 \text{ kJ/min}$$

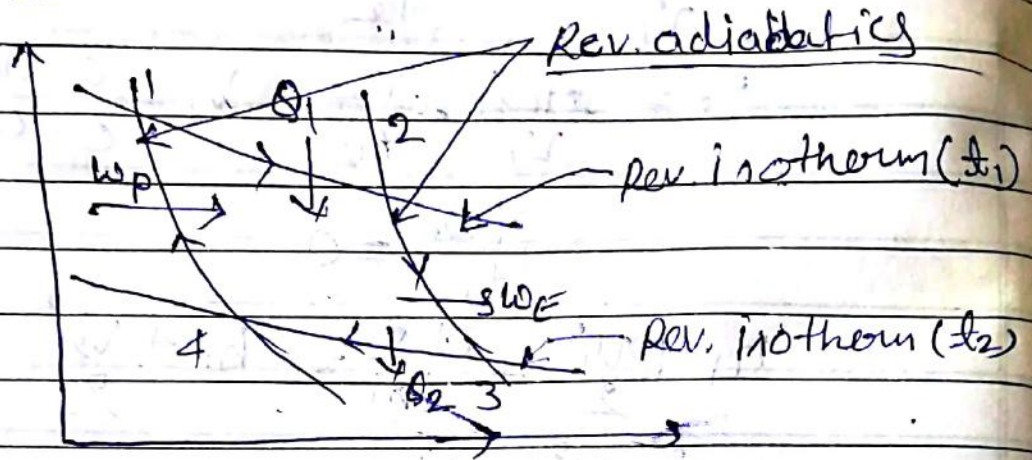
$$\therefore \eta_{th} = \frac{600}{2250} = 0.267 \text{ or } 26.7\%$$

$$Q_1 = W + Q_2$$

$$Q_2 = Q_1 - W = 2250 - 600$$

$$Q_2 = 1650 \text{ kJ/min}$$

Ans. 5: Carnot cycle:



$$\textcircled{1} \quad Q_1 = U_2 - U_1 + W_{1-2} \quad \text{---} \quad U_1 = U_2$$

$$\textcircled{2} \quad 0 = U_3 - U_2 + W_{2-3}$$

$$\textcircled{3} \quad -Q_2 = U_4 - U_3 - W_{3-4} \quad \text{---} \quad U_3 = U_4$$

$$\textcircled{4} \quad 0 = U_1 - U_4 - W_{4-1}$$

$$Q_1 - Q_2 = (W_{1-2} + W_{2-3}) - (W_{3-4} + W_{4-1})$$

$$\text{Or} \quad \sum_{\text{cycle}} Q_{\text{net}} = \sum_{\text{cycle}} W_{\text{net}}$$

Carnot theorem:-

It states that of all of heat engine operating b/w a given constant temp. source and a given constant temp. sink, none has a higher efficiency than a reversible engine.

## Sec-C

Ans: Given -  $T_1 = 273 + 15 = 308 \text{ K}$   
 $P_1 = 0.1 \text{ mpa} = 100 \text{ kN/m}^2$   
 $Q_1 = 2100 \text{ kJ/kg}$   
 $\gamma_k = 8, \quad \gamma = 1.4$

$$\eta_{\text{cycle}} = 1 - \frac{1}{\gamma^{\frac{\gamma-1}{\gamma}}} = 1 - \frac{1}{8^{0.4}}$$

$$\eta_{\text{cycle}} = 0.565 \text{ or } 56.5\%$$

We know that

$$\frac{T_2}{T_1} = \left(\frac{V_1}{V_2}\right)^{\gamma} \Rightarrow T_2 = \frac{T_1}{\gamma^{\frac{\gamma-1}{\gamma}}}$$

$$\therefore T_2 = \frac{308 \times 2.3}{8} = 708.4 \text{ K}$$

$$\frac{V_1}{V_2} = 8 = \frac{RT_1}{P_1 V_1} = \frac{0.287 \times 308}{100} = 0.844 \text{ m}^3/\text{kg}$$

$$V_2 = \frac{0.844}{8} = 0.11 \text{ m}^3/\text{kg}$$

$$Q_1 = c_v (T_3 - T_2) = 2100 \text{ kJ/kg}$$

$$T_3 - 708.4 = \frac{2100}{0.718} = 2925 \text{ K}$$

$$T_{\text{max}} = T_3 = 2925 + 708 = 3633 \text{ K}$$

$$\text{Now, } \left(\frac{P_2}{P_1}\right) = \left(\frac{V_1}{V_2}\right)^{\gamma} = (8)^{1.4} = 18.37$$

$$P_2 = 1.837 \text{ mpa}$$

$$\text{Again } \frac{P_2 V_2}{T_2} = \frac{P_3 V_3}{T_3}$$

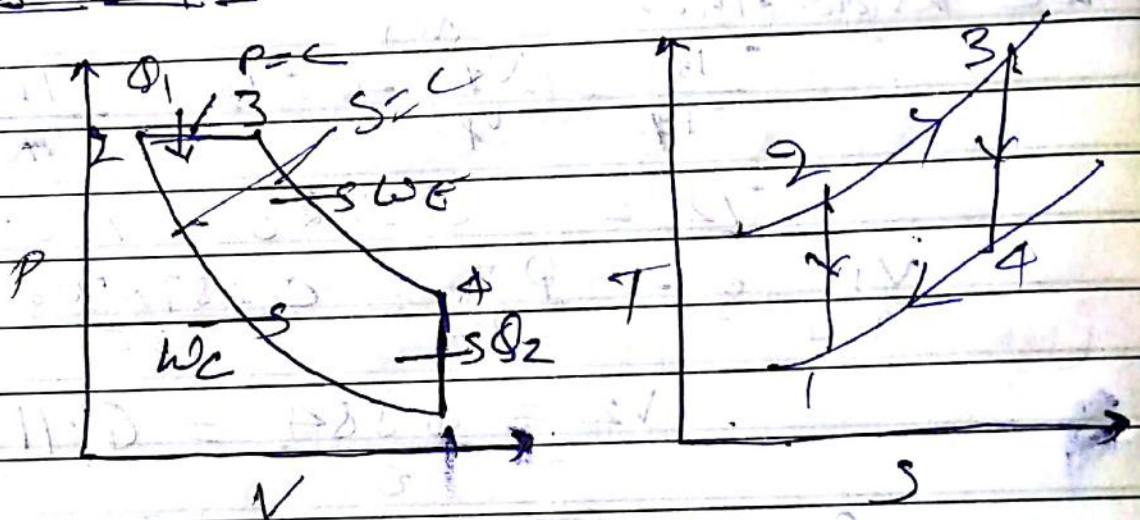
$$P_3 = \frac{P_2 V_2}{T_2} \times \frac{T_3}{V_3} \Rightarrow \frac{1.837 \times 363}{708} = 9.496$$

$$\begin{aligned} W_{net} &= Q_1 \times \eta_{cycle} = 2100 \times 0.565 \\ &= 1186.5 \text{ kJ/kg} \\ &= P_m (V_1 - V_2) = P_m (0.884 - 0.11) \end{aligned}$$

$$P_m = m \cdot e \cdot P = \frac{1186.5}{0.174}$$

$$= 1533 \text{ kPa} = 1.533 \text{ MPa}$$

Ans. 2 Diesel cycle



$$H.S \Rightarrow Q_1 = m c_p (T_3 - T_2)$$

$$H.R \Rightarrow Q_2 = m c_v (T_4 - T_1)$$

$$\eta = 1 - \frac{Q_2}{Q_1} = 1 - \frac{m c_v (T_4 - T_1)}{m c_p (T_3 - T_2)}$$

$$\eta_{cycle} = 1 - \frac{(T_4 - T_1)}{\gamma (T_3 - T_2)}$$

Now we have

$$\gamma_k = \frac{V_1}{V_2}, \quad \gamma_c = \frac{V_3}{V_2}, \quad \gamma_e = \frac{V_4}{V_3}$$

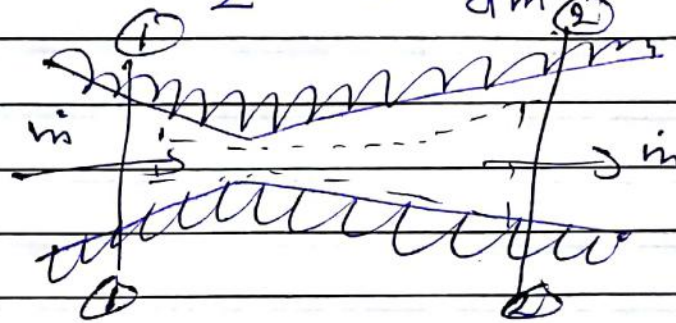
$$\gamma_k = \gamma_e \cdot \gamma_c$$

So we get value of  $T_1$  &  $T_2$ ,  $T_3$  &  $T_4$  by relation

$$\eta_{\text{diesel}} = 1 - \frac{1}{\gamma} \frac{1}{\gamma_c^{\frac{\gamma-1}{\gamma}}} \cdot \frac{\gamma_c^{\gamma} - 1}{\gamma_c - 1}$$

Ans 3. S.F.E.E for Nozzle:-

We know that,  $h_1 + \frac{V_1^2}{2} + Z_1 g + \frac{dQ}{dm} = h_2 + \frac{V_2^2}{2} + Z_2 g + \frac{dW}{dm}$



$$\frac{dQ}{dm} = 0$$

$$\frac{dW}{dm} = 0$$

$$h_1 + \frac{V_1^2}{2} = h_2 + \frac{V_2^2}{2}$$

$$h_1 = h_2 + \frac{V_2^2}{2}$$

$$V_2 = \sqrt{2(h_1 - h_2)} \text{ m/s}$$

(b) S.F.E.E for turbine

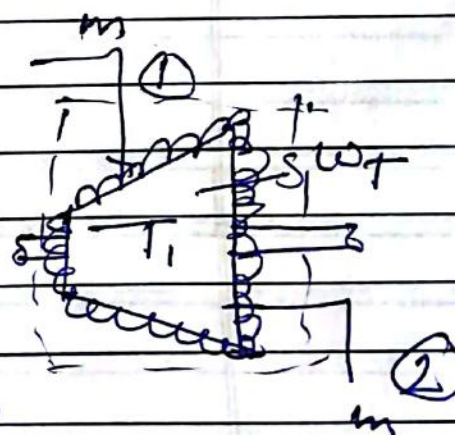
here,  $V_1 = 0$ ,  $\frac{dQ}{dm} = 0$ ,  $V_2 = 0$

so

S.F.E.E eq<sup>n</sup> for turbine is

~~$h_1 + \frac{V_1^2}{2}$~~

$$h_1 = h_2 + \frac{dW}{dm}$$



$$\frac{W_T}{m} = (h_1 - h_2)$$